



# Safety Data Sheet

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## Section 01 Identification

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<b>Product Identifier</b>	Citric Acid Anhydrous Citric Acid Anhydrous (30-80 Mesh), NSF® - 60 Citric Acid Anhydrous, Granular, NSF® - 60
<b>Other Means of Identification</b>	2-hydroxyl-1,2,3-propanyl-tri-carboxylic acid; CAS: 77-92-9
<b>Product Use and Restrictions on Use</b>	For commercial and industrial use. This product is NSF certified for use in drinking water, see section 15 and the NSF website for further information.
<b>Initial Supplier Identifier</b>	ClearTech Industries Inc. 1500 Quebec Avenue Saskatoon, SK. Canada S7K 1V7  Phone: 800.387.7503 Fax: 888.281.8109 <a href="http://www.cleartech.ca">www.cleartech.ca</a>
<b>Prepared By</b>	ClearTech Industries Inc. technical writer
<b>24-Hour Emergency Phone</b>	306.664.2522

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## Section 02 Hazard Identification

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### Physical Hazards

**Combustible dusts** Category 1

### Health Hazards

**Serious eye damage / eye irritation** Category 2

### Signal Word

**Warning**

### Hazard Statements

May form combustible dust concentrations in air.  
H319 Causes serious eye irritation.

### Pictograms



### Precautionary Statements

#### **Prevention**

P210 Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking.

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P264 Wash affected body parts thoroughly after handling.

P280 Wear , face protection

## Response

P305 P351 P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present  
P337 P313 and easy to do. Continue rinsing. If eye irritation persists: Get medical advice or attention.

## Hazards Not Otherwise Classified

Not available

## Supplemental Information

Not available

## Section 03 Composition / Information on Ingredients

### Hazardous Ingredients:

Chemical name	Common name(s)	CAS number	Concentration (w/w%)
2-hydroxypropane-1,2,3-tricarboxylic acid	Citric acid	77-92-9	>99.5%

## Section 04 First-Aid Measures

### Description of necessary first-aid measures

**Inhalation** Get medical advice / attention if you feel unwell or are concerned.

**Ingestion** Get medical advice / attention if you feel unwell or are concerned.

**Skin contact** Rinse skin with lukewarm, gently flowing water / shower for 5 minutes or until product is removed. If skin irritation occurs or if you feel unwell: Get medical advice / attention.

**Eye contact** Remove source of exposure or move person to fresh air. Rinse eyes cautiously with lukewarm, gently flowing water for several minutes, while holding the eyelids open. Remove contact lenses, if present and easy to do. Continue rinsing for 15 to 20 minutes. Take care not to rinse contaminated water into the unaffected eye or onto the face. If eye irritation persists: Get medical advice / attention.

### Most important symptoms and effects, both acute and delayed

**Inhalation** May cause respiratory irritation.

**Ingestion** May cause discomfort or nausea.

**Skin contact** Not irritating to skin.

**Eye contact** Causes serious eye irritation.

**Further information** For further information see Section 11 Toxicological Information.

## Section 05 Fire Fighting Measures

**Suitable extinguishing media** Extinguish fire using extinguishing agents suitable for the surrounding fire.

**Unsuitable extinguishing media** Water jets are not recommended in fires involving chemicals.

**Specific hazards arising from the chemical** May form combustible dust concentrations in air. Reacts with many metals to liberate hydrogen gas that can form explosive mixtures. In the event of a fire oxides of carbon may be released. Thermal decomposition occurs at 175 °C. Finely powdered citric acid anhydrous may form flammable or explosive mixtures with air at 500 mg of citric acid per litre.

**Special protective equipment for fire-fighters** Wear NIOSH-approved self-contained breathing apparatus and chemical-protective clothing.

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## Section 06 Accidental Release Measures

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<b>Personal Precautions / Protective Equipment / Emergency Procedures</b>	Wear appropriate personal protective equipment (See Section 08 Exposure Controls and Personal Protection). Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking. Stay upwind, ventilate area.
<b>Environmental Precautions</b>	Prevent material from entering waterways, sewers or confined spaces. Notify local health and wildlife officials. Notify operators of nearby water intakes.
<b>Methods and Materials for Containment and Cleaning Up</b>	Dry sweeping is not recommended. Pre-dampening the material or use of a vacuum is preferred. Shovel into clean, dry, labeled containers and cover. Flush area with water.

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## Section 07 Handling and Storage

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<b>Precautions for Safe Handling</b>	Use proper equipment for lifting and transporting all containers. Use sensible industrial hygiene and housekeeping practices. Wash thoroughly after handling. Avoid all situations that could lead to harmful exposure. Inspect containers for damage or leaks before handling. If the original label is damaged or missing replace with a workplace label. Have suitable emergency equipment for fires, spills and leaks readily available.
<b>Conditions for Safe Storage</b>	Store in a cool, dry, well-ventilated area, away from heat sources and incompatible materials. Always store in original labeled container. Keep containers tightly closed when not in use and when empty. Empty containers may contain hazardous residues. Protect label and keep it visible.
<b>Incompatibilities</b>	Bases, such as potassium hydroxide, sodium hydroxide, calcium hydroxide (slaked lime), ammonia, carbonates. Oxidizing agents, such as oxygen, hydrogen peroxide, sulphuric and nitric acids, hypochlorites and permanganates. Reducing agents, such as hydrogen, sodium borohydride, sulphur dioxide, thiosulphates, hydrazine, phosphites, carbon, and oxalic, formic and ascorbic acid. Metals, such as aluminum, copper, and zinc.

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## Section 08 Exposure Controls and Personal Protection

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### Exposure limits

There are no known exposure limits for this product.

### Engineering controls

<b>Ventilation Requirements</b>	Mechanical ventilation (dilution or local exhaust), process or personnel enclosure and control of process conditions should be provided in accordance with all fire codes and regulatory requirements. Supply sufficient replacement air to make up for air removed by exhaust systems.
<b>Other</b>	An eye wash bottle or eye wash station should be available, tested, and be in close proximity to the product being handled in accordance with provincial regulations.

### Protective equipment

The following are recommendations only. It is the responsibility of the employer / user to conduct a hazard assessment of the process in which this product being used and determine the proper engineering controls and PPE for their process. Additional regulatory and safety information should be sought from local authorities and, if needed, a professional industrial hygienist.

<b>Eye and face protection</b>	Where there is potential eye or face exposure, tightly fitting chemical goggles are recommended. Contact lenses are not recommended; they may contribute to severe eye injury.
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<b>Hand and body protection</b>	Where handling this product it is recommended that skin contact is avoided.
<b>Respiratory protection</b>	In case of insufficient ventilation wear suitable respiratory equipment.
<b>Thermal hazards</b>	Not available

## Section 09 Physical and Chemical Properties

### Appearance

<b>Physical state</b>	Solid
<b>Colour</b>	White
<b>Odour</b>	Odourless
<b>Odour threshold</b>	Not applicable

### Property

<b>pH</b>	2.2 @ 1%
<b>Melting point / freezing point</b>	153 °C
<b>Initial boiling point and boiling range</b>	Decomposes
<b>Flash point</b>	Not applicable
<b>Evaporation rate</b>	Not available
<b>Flammability</b>	Finely powdered citric acid anhydrous may form flammable or explosive mixtures with air at 500 mg of citric acid per litre.
<b>Upper flammable limit</b>	Not available
<b>Lower flammable limit</b>	Not available
<b>Vapour pressure</b>	$2.21 \times 10^{-6}$ Pa
<b>Vapour density</b>	Not available
<b>Relative density</b>	1.54-1.665 g/cm <sup>3</sup>
<b>Solubility</b>	590 g/L @ 20 °C in water
<b>Partition coefficient: n-octanol/water</b>	Log Kow: -0.2 to -1.8
<b>Auto-ignition temperature</b>	1010 °C
<b>Decomposition temperature</b>	175 °C
<b>Viscosity</b>	Not applicable
<b>Specific gravity</b>	Not applicable
<b>Particle characteristics</b>	Particle size: For 30-80 Mesh: 95% of particles are 125-595 microns Particle shape: Crystals or granules
<b>Formula</b>	C <sub>6</sub> H <sub>8</sub> O <sub>7</sub>
<b>Molecular weight</b>	192.027 g/mol

## Section 10 Stability and Reactivity

<b>Reactivity</b>	Reacts with many metals to liberate hydrogen gas that can form explosive mixtures. Reacts violently with bases.
<b>Stability</b>	This product is stable if stored according to the recommendations in Section 07.
<b>Possibility of hazardous reactions</b>	Hazardous polymerization is not known to occur.
<b>Conditions to avoid</b>	Avoid contact with incompatible materials. Do not heat.

<b>Incompatible materials</b>	<p>Bases, such as potassium hydroxide, sodium hydroxide, calcium hydroxide (slaked lime), ammonia, carbonates.</p> <p>Oxidizing agents, such as oxygen, hydrogen peroxide, sulphuric and nitric acids, hypochlorites and permanganates.</p> <p>Reducing agents, such as hydrogen, sodium borohydride, sulphur dioxide, thiosulphates, hydrazine, phosphites, carbon, and oxalic, formic and ascorbic acid.</p> <p>Metals, such as aluminum, copper, and zinc.</p>
<b>Hazardous decomposition products</b>	Thermal decomposition may produce oxides of carbon. Thermal decomposition occurs at 175 °C.

## Section 11 Toxicological Information

### Acute Toxicity (LD50 / LC50 values)

Component	Route	Species	Value	Exposure time
citric acid	Oral	mouse	5400 mg/kg	
	Dermal	rat	>2000 mg/kg	24 hours

### Toxic Health Effect Summary

<b>Chemical characteristics</b>	Citric acid is a metabolic intermediate vital to the TCA respiration pathway found in all animal and plant cells. There is little evidence that citric acid and the citrate salts have deleterious effects, even in large doses. Indeed there is some support for the fact that citric acid in the human diet is favourable by inhibiting the formation of calcium oxalate kidney and bladder stones. This statement is applicable to the citrate salts since once absorbed citrate salts will dissociate into citric acid and their counter-ion.
<b>Skin</b>	Not irritating to skin. Not irritating to skin.
<b>Ingestion</b>	May cause discomfort or nausea.
<b>Inhalation</b>	May cause respiratory irritation.
<b>Eye contact</b>	Causes serious eye irritation.
<b>Sensitization</b>	This product and its components at their listed concentration have no known sensitizing effects.
<b>Mutagenicity</b>	This product and its components at their listed concentration have no known mutagenic effects.
<b>Carcinogenicity</b>	This product and its components at their listed concentration have no known carcinogenic effects.
<b>Reproductive toxicity</b>	This product and its components at their listed concentration have no known reproductive effects.
<b>Specific organ toxicity</b>	This product and its components at their listed concentration have no known effects on specific organs.
<b>Aspiration hazard</b>	Not available
<b>Synergistic materials</b>	Not available

## Section 12 Ecological Information

### Ecotoxicity

Component	Type	Species	Value	Exposure Time
Citric acid	LC50	Leuciscus idus melanotus	440 mg/L	48 hours
	EC50	Daphnia magna	1535 mg/L	24 hours

**Biodegradability** The domestic substance list categorizes citric acid as non-persistent.

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<b>Bioaccumulation</b>	The domestic substance list categorizes citric acid as non-bioaccumulative.
<b>Mobility</b>	This product is water soluble, is not predicted to adsorb to soil and may contaminate ground water.
<b>Other adverse effects</b>	Not available

## Section 13 Disposal Considerations

<b>Waste From Residues / Unused Products</b>	Dispose in accordance with all federal, provincial, and local regulations including the Canadian Environmental Protection Act.
<b>Contaminated Packaging</b>	Do not remove label, follow label warnings even after the container is empty. Empty containers should be recycled or disposed of at an approved waste handling facility.

## Section 14 Transport Information

<b>UN number</b>	This product does not meet the definition of dangerous goods per Part 2 of Transport of Dangerous Goods Regulations.
<b>UN proper shipping name and description</b>	Not available
<b>Transport hazard class(es)</b>	Not available
<b>Packing group</b>	Not available
<b>Excepted quantities</b>	Not available
<b>Environmental hazards</b>	Not listed as a marine pollutant under Canadian TDG Regulations, schedule III.
<b>Special precautions</b>	No special provisions
<b>Transport in bulk</b>	ERAP index: not available
<b>Additional information</b>	MARPOL 73/78 and IBC Code: This product is not listed in Chapter 17 of the IBC Code. Secure containers (full or empty) during shipment and ensure all caps, valves, or closures are secured in the closed position.

**TDG PRODUCT CLASSIFICATION:** This product has been classified on the preparation date specified at section 16 of this SDS, for transportation in accordance with the requirements of part 2 of the Transportation of Dangerous Goods Regulations. If applicable, testing and published test data regarding the classification of this product are listed in the references at section 16 of this SDS.

## Section 15 Regulatory Information.

**NOTE: THE PRODUCT LISTED ON THIS SAFETY DATA SHEET HAS BEEN CLASSIFIED IN ACCORDANCE WITH THE HAZARD CRITERIA OF THE CANADIAN HAZARDOUS PRODUCTS REGULATIONS. THIS SAFETY DATA SHEET CONTAINS ALL INFORMATION REQUIRED BY THOSE REGULATIONS.**

All components of this product appear on the domestic substance list.

NSF Certification: Citric Acid Anhydrous (30-80 Mesh), NSF® - 60 and Citric Acid Anhydrous, Granular, NSF® - 60 are certified under NSF / ANSI Standard 60 for membrane cleaning and pH adjustment at a maximum dosage of: 125 mg/L. NSF product use restrictions based on requirements obtained from the NSF website; consult NSF website for current requirements.

## Section 16 Other Information

**Date of latest revision: January 19, 2023**

**Note:** The responsibility to provide a safe workplace remains with the buyer / user. The buyer / user should consider the health hazards and safety information contained herein as a guide and should take those precautions required in an individual operation to instruct employees and develop work practice procedures for a safe work environment. The information contained herein is, to the best of our knowledge and belief, accurate. However, since the conditions of handling and use are beyond our control, we make no guarantee of results, and assume no liability for damages incurred by the use of this material. It is the responsibility of the buyer / user to comply with all applicable laws and regulations regarding handling, using, reselling and shipping this product.

**Attention: Receiver of the chemical goods / SDS coordinator**

As part of our commitment to the RDC Responsible Distribution® initiative, ClearTech Industries Inc. and its associated companies require, as a condition of sale, that you forward the attached Safety Data Sheet(s) to all affected employees, customers, and end-users. ClearTech will send any available supplementary handling, health, and safety information to you at your request.

If you have any questions or concerns please call our customer service center.

**References:**

- 1) NIOSH Pocket Guide to Chemical Hazards
- 2) WorkSafe BC E-Limit
- 3) ECHA - Registered Substance Dossier
- 4) Transportation of Dangerous Goods Canada
- 5) Globally Harmonized System of Classification and Labelling of Chemicals (GHS) *Seventh revised edition*
- 6) International Code for the Construction and Equipment of Ships Carrying Dangerous Chemicals in Bulk (IBC Code) 2007 Edition
- 7) The ACS Style Guide